

## Answer Sheets for CHE654 Homework Set #5 (100 Points)

**Note:** For all problems, submit a copy of your process flow diagram and a copy of your input summary of the process.

### 36. Extracting Compound Y via an Evaporator and a Stripper Using Compound X (25 points)

Answer the following questions:

Required heat transfer area of the evaporator = \_\_\_\_\_ ft<sup>2</sup>

Concentration of Compound Y in the liquor stream from the evaporator = \_\_\_\_\_ mol%

Concentration of Compound Y in the outlet liquid stream of the stripper = \_\_\_\_\_ mol%

### 39. Property Requirements and PCES, II (25 points)

(a) Required parameters:

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(b) Values of estimated parameters:

MW = \_\_\_\_\_;  $T_C$  = \_\_\_\_\_ K;  $P_C$  = \_\_\_\_\_ N/m<sup>2</sup>

$Z_C$  = \_\_\_\_\_; CPIG at 300 K = \_\_\_\_\_ J/kmol-K

DHFORM = \_\_\_\_\_ J/kmol; DGFORM = \_\_\_\_\_ J/kmol

OMEGA = \_\_\_\_\_; DHVLB = \_\_\_\_\_ J/kmol

VB = \_\_\_\_\_ m<sup>3</sup>/kmol

(c)  $H_V^{IG}$  = \_\_\_\_\_ Btu/lbmol

$H_L$  = \_\_\_\_\_ Btu/lbmol

(d) Tdew at 1 atm = \_\_\_\_\_ °C

Tdew at 50 atm = \_\_\_\_\_ °C

**41. Recovering Acetone from a Wastewater Stream (25 points)**

Answer the following questions:

D:F molar ratio in Column 1 = \_\_\_\_\_

Purity of acetone product stream = \_\_\_\_\_ mole%

**42. Extractive Distillation, II (25 points)**

Answer the following questions:

Mole % purity of butene in the distillate of the extractive column = \_\_\_\_\_

Mole % purity of butadiene in the distillate of the solvent recovery column = \_\_\_\_\_