

## Answer Sheet for CHE494 Homework Set #3 (100 points)

**Note:** For all problems, submit a copy of your process flow diagram and a copy of your input summary of the process.

### 11. (30 points) *Using ASPEN PLUS to Perform Simple Calculations*

Answer the following questions:

(a) (i) At  $P = 1.01325$  bar (1 atm):

Bubble point temperature of the mixture = \_\_\_\_\_ °C

Dew point temperature of the mixture = \_\_\_\_\_ °C

(ii) At  $T = 100$  °C:

Bubble point pressure of the mixture = \_\_\_\_\_ bar

Dew point pressure of the mixture = \_\_\_\_\_ bar

(iii) Temperature at which the flash will produce a vapor stream containing

exactly 50 mol% acetone = \_\_\_\_\_ °C

(b) Composition (mass fractions) of the benzene-toluene feed = \_\_\_\_\_

\_\_\_\_\_

### 12. (30 points) *Simulation of a Cyclohexane Production Process*

Answer the following questions:

1. Pressure of the column condenser = \_\_\_\_\_ psia

2. Purge fraction = \_\_\_\_\_

3. Temperature of the flash vessel = \_\_\_\_\_ °F

4. Purity of cyclohexane (mole%) in the product stream = \_\_\_\_\_ %

**13. (40 points) *Simulating an Acetone Production Process***

Answer the following questions:

1.  $\Delta P$  across the reactor = \_\_\_\_\_ psia
2. % conversion of the reaction based on IPA = \_\_\_\_\_ %
3. Temperature of the condenser in Column-1 = \_\_\_\_\_ °F
4. Product purity (mole%) of acetone in liquid distillate of Column-1 = \_\_\_\_\_ %